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Freezing point depression of solution

The study of depression in freezing point of a solution is called. Which of the following solutions will have higher freezing point depression and why. Which of the following solution shows maximum depression in freezing point. The freezing point depression of 0.1 molar solution of acetic acid. During depression of freezing point in a solution the following are in equilibrium. The depression in freezing point of 0.1m aqueous solution of hcl csu_4 . The depression in freezing point of 0.01 m aqueous solution of urea. Which of the following statements correctly describe freezing point depression for a solution.

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